1. COURSE DECRIPTION - GENERAL	NFORMATION				
1.1 Course teacher	Assoc. Prof. Roberta Petlevski, PhD Prof. József Petrik, PhD	1.6. Year of study	3 rd		
1.2. Name of the course	General Clinical Biochemistry	1.7. Credit value (ECTS)	13,5		
1.3. Associate teachers	Associate Professor Dunja Rogić, PhD Anita Somborac Bačura, PhD Marija Grdić Rajković, PhD Andrea Hulina, mag. med. biochem.	1.8. Type of instruction (number of hours L+E+S+e-learning)	60+90+30		
1.4. Study programme (undergraduate, graduate, integrated)	Integrated stud of Medical biochemistry	1.9. Expected enrolment in the course	20-25		
1.5. Status of the course	Compulsory	1.10. Level of use of e-learning (1, 2, 3 level), percentage of instruction in the course on line (20% maximum)	2 nd		
2. COURSE DESCRIPTION		,			
2.4. Course shipatives	Introducing the students to analytical quality specification of methods used for the measurement of analytes in biological				
2.1. Course objectives	fluids, cells and tissues in order to assess the pathological changes in the human organism.				
2.2. Enrolment requirements and required entry competences for the course	Enrolment requirements for this subject: Biochemistry and Physiology and Human Anatomy courses completed				
	Conditions for taking an examination in this subject: passed examination in Biochemistry and Physiology and Human				
	Anatomy.				
2.3. Learning outcomes at the level of the study programme to which the course contributes	Applying expert knowledge and skills in making and analytical interpretation of laboratory tests				
	The implementation of quality control systems in such a way that the rules of good laboratory practice				
	Improving laboratory diagnostics in the field of general medical biochemistry.				
	After passing the course the student will/will be able to:				
2.4. Expected learning outcomes at the level of the course (4-10 learning outcomes)	Perform laboratory method for the determination of each analyte in a clinical specimen;				
	2. To assess the clinical significance of certain biochemical parameters;				
	3. Anticipate and avoid possible errors that can occur in preanalytical, analytical and postanalytical phase of laboratory				
	work;				
	4. Describe the biochemical mechanisms that alter the concentration of a particular analytes in the body;				
	5. Explain the results of laboratory tests and associate them with certain pathological conditions;				
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	6. Identify the diagnostic poss	sibilities of	using molecular biologica	I methods in the d	etection of pathological condit	ions.
2.5. Course content broken down in detail by weekly class schedule (syllabus)	LECTURES AND SEMINARS: Principles of medical biod techniques and instrumenta Water and electrolytes: wa chloride, lithium, calcium, p Acid-base balance and bl kidney and lung as control Proteins in body fluids: pro and catabolism of protein compounds: amino acids, u Carbohydrates: metabolish hypoglicaemia, glycated hamethods of determination. Lipids and lypoproteins: metabolish determination. Lipids and lypoproteins: metabolish serum enzyme concentrations, LDH, AST, Trace elements: iron, coppof determination. Urin: composition, format quantitative urin analysis, u Vitamins: solubility, health of the synthesis: porphyriae	chemistry la ations, qua ter and ele phosphates lood gases systems, a oteins in p is, individu urea, creatir am and h aemoglobir etabolism a hospholipic is, organ-s ALT, ALP, per, zinc, s tion of uri urin sedime effects, def s, deficien Hemoglob	aboratory work: preanaly ality assurance, methods, ectrolytes balance and regist, magnesium, disturbances; definitions and diagnoracid-base disturbances, malasma and serum, urine, al plasma proteins, methormonal regulation, don, fructosamine, disorders and hormonal regulation, dos, disturbances, methods, disturbances, methods, elenium, characteristics one, excretion, filtration, ent, disorders, faeces and ficiencies, methods of detoy in the enzymes of the pin and hem disintegration.	rtical, analytical a reference intervals gulations, fluid con es and disorders, lastic parameters of aethods of determinationium, disorders, labetes mellitus, in the metabolism structure and physis of determination eymes, methods lipase, principles of essential-trace measurement of disorders. The porphyrin pathon: hemoglobin values and porphyrin pathon: hemoglobin values and disorders.	and postanalytical phases, sp s. Impartments, water, sodium, postanation. In acid-base balance, buffer nation. In acid-base balance, buffer nat	ecimens, otassium, systems, etabolism nitrogen olications, glycogen, oproteins, catalytic methods etive and liagnosis,
2.6. Type of instruction	lectures seminars and workshops exercises online in entirety mixed e- learning field work		independent study multimedia and the internet laboratory work with the mentor (other)		2.7. Comments:	
2.8. Student responsibilities	Regular attendance of lectures, exercises and seminars.					
2.9. Screening of student's work (specify	Class attendance		Research	l l	Practical training	·

the proportion of ECTS credits for each activity so that the total number of CTS credits is equal to the credit value of the course)	Experimental work		Report		(Othe	erdescribe)	
	Essay		Seminar essay	0.5	(Othe	erdescribe)	
	Tests		Oral exam	10.5	(Othe	(Otherdescribe)	
	Written exam	3	Project		(Othe	erdescribe)	
2.10. Grading and evaluation of student work over the course of instruction and at a final exam	Teaching methods: lectures, exercise, seminars. Methods of evaluation and assessment of learning outcomes: written exam, oral exam, entrance and final colloquium for practical classes.						
2.11. Required literature (available at the library and via other media)			Title			Number of copies at the library	Availability via other media
	Štrausova Medicinska biokemija – editor: Medicinska naklada 2009.				18		
	Lecture handouts						Available on the website
	Laboratory exercises (internal	script)					
2.12. Optional literature						•	•
2.13. Methods of monitoring quality that ensure acquisition of exit competences	Outcomes 2 – 6 are tested by outcome 1 is tested by the final			utcome 1 during	g exercise	es in the laboratory	y, while